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The Iodine clock occurred when sev​r​ox​i​da​tion-re​duc​tion re​ac​tions were completed with accuracy of a clock. The rate of the reaction of interest between iodide I-​ ​ and persulfate ions S​2O​ and

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A ferriin solution was mixed with Sulfuric acidH​2S​O​4​ to separate the ferrous ions Fe2​ ​ and the rate of dissociation of the phenanthroline ligand was observed. Plots were used and are provided to determine the appropriate rate order the rate law and k value. After determining the k values at ...

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​Titration Data for 25.00 mL of the Unknown Solution is given and tabulated. Titration curve with product name is shown. Sample calculations written step by step for Part 1: A Mysterious Acid and Part 2: Polyprotic Acids. All post lab questions correct and included.

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